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E1-26 The room volume is  $(21.13 \text{ m})(3.0:3048 \text{ m})(3 = 92:8\text{m}^3$ . The mass contained in the room is  $(92:8\text{m}^3)(1:21\text{kg}=\text{m}^3) = 112\text{kg}$ : E1-27 One mole of sugar cubes would have a volume of  $N \text{ A } 1:0 \text{ cm}^3$ , where  $N \text{ A}$  is the Avogadro constant. Since the volume of a cube is equal to the length cubed,  $V = l^3$ , then  $l = \sqrt[3]{V}$   $N \text{ A } \text{ cm} = 8:4 \text{ } 107 \text{ cm}$ . E1-28 The number of seconds in a week is

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