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Steps for Writing Lewis Structures. Example \(\PageIndex{2}\}) 1. Determine the total number of valence electrons in the molecule or ion. Each hydrogen atom (group 1) has one valence electron, carbon (group 14) has 4 valence electrons, and oxygen (group 16) has 6 valence electrons, for a total of $(2)(1) + 4 + 6 = 12$ valence electrons. 2.

10.5: Writing Lewis Structures for Covalent Compounds ...

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Information: Steps for Drawing Lewis Structures for Covalent Compounds Study the two examples in the table of how to write structures for CO3 2-and NH3. Make sure you understand each of the five steps. CO3 2-NH 3 Step #1: Add up the number of valence electrons that should be included in the Lewis Structure. 4 + 3(6) + 2 = 24 (carbon has four ...

ChemQuest 23 : Drawing Covalent Compounds

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14 kt. gold (mixture of silver and gold) 1 1 Homogeneous Critical Thinking Questions 16. What is the difference between a "phase of matter" and a "state of matter"? Define each term as best you can. The "state" of matter tells us whether the substance is a solid, liquid, or gas. A "phase" of matter is a section of matter that is the ...